Exam. Code: 103203 Subject Code: 1348

B.A./B.Sc. 3rd Semester CHEMISTRY (Organic Chemistry—II)

Time Allowed—3 Hours]

[Maximum Marks—35

PART—A

Note:—ALL questions are compulsory. Each question carries 1 mark.

1. Assign E/Z configurations to the following:

$$H_3C$$
 CHO
$$H_2C = H$$

(B)
$$(H_3C)_2HC \xrightarrow{H} C_6H_5$$

$$C_2H_5$$

- 2. Draw the structure of diene: (2Z, 4Z)-1-chloro-2, 4-heptadiene.
- 3. Assign R/S configurations to the following:

CHO
$$C_2H_5$$

(A) $HOH_2C - H$ (B) $H - CI$
 CH_3 C_6H_5

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- 4. Acid catalysed dehydration of tert-butyl alcohol occurs faster than n-butyl alcohol, why?
- 5. Phenol is insoluble in aqueous sodium carbonate solution while 2, 4, 6-trinitrophenol is soluble, why?
- 6. Alkylation of 2, 6-di-tert-butyl phenol yields C-alkylated as major product over O-alkylation, why?
- 7. Aromatic aldehydes undergo nucleophilic addition reactions less readily than aliphatic aldehydes, explain.
- 8. Why CN is required for benzoin condensation?

 $8 \times 1 = 8$

PART—B

Note :— Attempt **TWO** questions from each section. All questions carry equal marks.

SECTION—I

- 9. Write down various conformations of n-butane and mark them on energy profile diagram. $4\frac{1}{2}$
- Draw Newman projection formula of chair and boat form of cyclohexane and comment upon their relative stability.
- 11. "Walden Inversion does not necessarily mean optical inversion". Justify the statement. 4½

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12. Identify the products in the following reaction and provide suitable mechanism: 4½

$$\begin{array}{c|c}
OH & H_2SO_4 \\
CH_3 & A+B
\end{array}$$

13. Predict the product in the following reaction and provide a suitable mechanism for it: 4½

$$\begin{array}{c|c} & & \\ & &$$

14. How will you convert anisole to p-anisaldehyde?

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SECTION—III

15. Predict the product in the following reaction and write a suitable mechanism: 4½

$$C_6H_5$$
 H
 C_6H_5
 H
 C_6H_5
 H
 C_6H_5
 H
 C_6H_5
 C_6

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- 16. How will you convert propiophenone to n-propyl benzene? Give a suitable mechanism. 4½
- 17. Predict the reagents required for the following transformation along with a suitable mechanism:

$$\bigcirc = 0 \longrightarrow \bigcirc = CHCOOH$$

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